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Cytotoxic and antioxidant activities of diterpenes and sterols from the Vietnamese soft coral *Lobophytum compactum*

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ABSTRACT

Two new diterpenes, lobocompactols A (1) and B (2), and five known compounds (3–7) were isolated from the methanol extract of the soft coral *Lobophytum compactum* using combined chromatographic methods and identified based on NMR and MS data. Each compound was evaluated for cytotoxic activity against A549 (lung) and HL-60 (acute promyelocytic leukemia) human cancer cell lines. Among them, compound 5 exhibited strong cytotoxic activity against the A549 cell line with an IC₅₀ of 4.97 \pm 0.06 μ M. Compounds 3, 4, and 7 showed moderate activity with IC₅₀ values of 23.03 \pm 0.76, 31.13 \pm 0.08, and 36.45 \pm 0.01 μ M, respectively. The cytotoxicity of 5 on the A549 cells was comparable to that of the positive control, mitoxantrone (MX). All compounds exhibited moderate cytotoxicity against the HL-60 cell line, with IC₅₀ values ranging from 17.80 \pm 1.43 to 59.06 \pm 2.31 μ M. Their antioxidant activity was also measured using oxygen radical absorbance capacity method, compounds 1 and 2 exhibiting moderate peroxyl radical scavenging activity of 1.4 and 1.3 μ M Trolox equivalents, respectively, at a concentration of 5 μ M.

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Marine organisms are rich in biologically active compounds, many of which have unique structures that are not found in terrestrial organisms. In particular, soft corals have received a great deal of attention from scientists worldwide because of the ease with which samples can be obtained and the diversity of chemical constituents and biological activities. Soft corals comprise a significant group of marine organisms and occur in coral reefs worldwide, with a marked preference for tropical waters at depths between 5 and 30 m. The genera Sinularia, Lobophytum and Sarcophyton are the most prolific. Interestingly, some Sarcophyton species contain diterpenes at levels of up to 10% of their dry weight. Such large amounts of these secondary metabolites may be important to the survival of octocorals with regard to defensive, competitive, reproductive, and possibly pheromonal functions. It is conceivable that soft corals, which lack any form of physical defense, are protected from predation by the sheer levels of diterpene toxins in their tissue.2

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As a part of an ongoing project to catalogue the chemical constituents and biological activities of Vietnamese marine organisms, we report herein the isolation, structural elucidation, and evaluation of *in vitro* cytotoxic and antioxidant activities of four diterpenes including two new compounds, lobocompactols A (1) and B (2), and three sterols from the methanol extract of the soft coral *Lobophytum compactum* (see Fig. 1).

The specimens of *Lobophytum compactum* were collected in Bay Canh island, Truong Sa archipelago, Khanh Hoa, Viet Nam during January 2010 and deep frozen until used. The sample was identified by one of us, Dr. Do Cong Thung. A voucher of specimen (No. NCCB M-14) was deposited at Institute of Marine Biochemistry and Institute of Marine Resources and Environment, VAST, Viet Nam.

Fresh frozen samples of the soft coral L. compactum (30 kg) were finely ground and extracted three times with hot MeOH (50 °C for 3 h each time) and concentrated under reduced pressure to yield the MeOH extract (210 g). This extract was suspended in water (2 L) and partitioned in turn with chloroform (3 \times 2 L). The chloroform extract (140 g) was crudely separated on a silica gel chromatographic column (CC) with a gradient elution of ethyl acetate in n-hexane from 0% to 100% to yield five fractions (LC1–LC5). Fraction LC1 (47.1 g) was further separated on a silica gel CC eluting

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Figure 1. Structures of 1-7.

with n-hexane–acetone (30:1, v/v) to obtain nine sub-fractions, LC1A–LC1I. Purification of sub-fraction LC1C (2.5 g) on a silica gel CC with n-hexane–ethyl acetate (25:1, v/v) furnished lobatriene (**4**, 200 mg)^{3,4} as a colorless oil. Fraction LC2 (32.2 g) was divided into seven sub-fractions, LC2A–LC2G, by a silica gel CC eluted with chloroform–acetone (18:1, v/v). Sub-fraction LC2D (3.5 g) was further separated on an YMC CC using an acetone–water (5:1, v/v) as eluent to give lobocompactol A (**1**, 14 mg),⁵ lobocompactol B (**2**, 23 mg),⁵ and lobatrienolide (**3**, 38 mg)⁶ as colorless oils. Fraction LC4 (13.2 g) was further separated by a silica gel CC eluted with n-hexane–acetone (3:1, v/v) to obtain five sub-fractions,

Table 1
The NMR spectroscopic data of 1 and 2

Position		1		2
	$\delta_{C}^{a,b}$	$\delta_{H}^{a,c}$ mult. (J, Hz)	$\delta_{C}^{a,b}$	$\delta_{\rm H}{}^{\rm a,c}$ mult. (J, Hz)
1	125.4	5.36 d (1.0)	125.4	5.36 br s
2	25.3	2.13 m/2.39 m	24.5	2.23 m/2.40 m
3	25.3	2.13 m	28.8	1.30 m/1.66 m
4	149.1	_	149.2	_
5	75.9	3.88 br d (8.5)	75.5	4.05 br d (3.5)
6	36.5	1.40 m	32.5	1.38 m/1.80 m
7	39.5	1.67 m	37.1	2.12 m
8	31.5	1.42 m	31.9	1.37 m
9	40.2	1.94 m/2.14 m	39.7	1.95 m/2.12 m
10	133.5	_	132.7	_
11	15.7	1.64 s	15.5	1.62 s
12	115.4	5.07 br s/5.10 br s	109.7	5.13 br s/5.28 br s
13	140.5	_	145.1	_
14	67.6	4.12 br d (16.0)	66.9	4.22 br d (16.0)
		4.25 br d (16.0)		4.31 br d (16.0)
15	117.8	5.54 m	117.7	5.66 m
16	25.3	1.93 m (α)	25.1	1.97 m (α)
		2.13 m (β)		2.15 m (β)
17	80.3	3.25 dd (3.0, 11.0)	79.9	3.27 dd (3.0, 11.0)
18	71.7	_	71.6	_
19	26.1	1.21 s	26.2	1.21 s
20	23.8	1.16 s	23.5	1.15 s

All assignments were done by HSQC, HMBC, COSY, and ROESY experiments.

LC4A–LC4E. Compounds 3β ,11-dihydroxy-24-methylene-9,11-secocholestan-5-en-9-one (**5**, 23 mg), 7 (24S)-ergostane- 3β ,5 α ,- 6β ,25-tetraol (**6**, 21 mg), 8 and (24S)-ergostane- 3β ,5 α ,6 β ,25-tetraol 25-monoacetate (**7**, 37 mg) 9 were purified as white crystals from sub-fraction LC4C (5 g) following a two-stage separation beginning with a silica gel CC eluted with chloroform-acetone (7:1, v/v), followed by an YMC CC eluted with acetone–water (4:1, v/v).

Lobocompactol A (1) was isolated as a colorless oil with a molecular formula of C₂₀H₃₂O₃ as determined by ion peaks in the ESIMS at m/z 303 $[M-H_2O+H]^+$ (positive) and in the Fourier transform ion cyclotron resonance mass spectrum (FTICRMS) at m/z $343.22462 \text{ [M+Na]}^+ \text{ (calcd for } C_{20}H_{32}O_3Na, 343.22491). The {}^1H$ NMR spectrum of **1** revealed four olefinic [δ 5.07 (1H, br s), 5.10 (1H, br s), 5.36 (1H, d, I = 1.0 Hz), and 5.54 (1H, m)] and four oxygenated protons [δ 3.25 (1H, dd, J = 3.0 and 11.0 Hz), 3.88 (1H, br d, I = 8.5 Hz), 4.12 (1H, br d, I = 16.0 Hz), and 4.25 (1H, br d, I = 16.0 Hz)]. Moreover, the proton signals at δ 1.16, 1.21, and 1.64 (each 3H, s) indicated the presence of three tertiary methyl groups. The ¹³C NMR spectrum indicated 20 carbons including typical signals of three methyl (δ 15.7, 23.8, and 26.1), one oxymethylene (δ 67.6), two oxymethine (δ 75.9 and 80.3), and one oxygenated quaternary carbons (δ 71.7), detected by DEPT experiments. In addition, the presence of one olefinic methylene (δ 115.4), two olefinic methine (δ 117.8 and 125.4), and three quaternary olefinic carbons (δ 133.5, 140.5, and 149.1) indicated three double bonds. All of the protons were assigned to relevant carbons by HSQC correlations (Table 1). Compound 1 was considered to be a bicyclic diterpene by its two remaining degrees of unsaturation

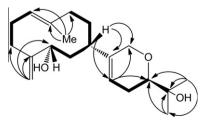


Figure 2. Selected HMBC (\rightarrow) and $^{1}H-^{1}H$ COSY (-) correlations of 1.

^a Measured in CDCl₃.

^b 125 MHz.

c 500 MHz.

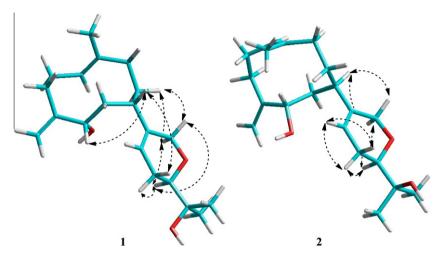


Figure 3. Important ROESY correlations of 1 and 2.

Table 2Effects of **1–7** on the growth of human cancer cells

Compound	IC_{50}^{a} (μM)		
	A549 (Lung)	HL-60 (Leukemia)	
1	>100	48.24 ± 1.33	
2	>100	37.51 ± 0.38	
3	23.03 ± 0.76	24.79 ± 0.77	
4	31.13 ± 0.08	33.82 ± 0.27	
5	4.97 ± 0.06	17.80 ± 1.43	
6	>100	59.06 ± 2.31	
7	36.45 ± 0.01	42.76 ± 2.85	
MX^b	7.83 ± 0.04	8.10 ± 0.60	

 $[^]a$ IC $_{50}$ (concentration that inhibits 50% of cell growth). Compounds were tested at a maximum concentration of 100 $\mu M.$ Data are presented as the mean \pm standard deviation (SD) of experiments performed in triplicate.

(five in total from the molecular formula minus three double bonds). The ¹H-¹H COSY experiment of **1** allowed the assignments

of the proton–proton correlations for $H-1/H_2-2/H_2-3$, $H-5/H_2-6/H-7/H_2-8/H_2-9$, and $H-15/H_2-16/H-17$. These data together with the HMBC cross peaks between H_3-11 and C-1/C-9/C-10 and between H_2-12 and C-3/C-4/C-5 confirmed the connectivities from C-1 to C-10, indicating a ten-membered ring and the locations of the exomethylene, olefinic methyl, and hydroxy groups (see Fig. 2).

The configuration of the C-1/C-10 double bond was identified as *trans* based on the ^{13}C NMR chemical shift of C-11 at δ 15.7 (<20 ppm). On the other hand, the chemical shifts of C-3 (δ 25.3), C-4 (δ 149.1), C-5 (δ 75.9), and C-6 (δ 36.5) of **1** suggested hydroxyl group at C-5 to be α -orientation by agreement of them with the corresponding data of the gyrosanol C¹¹ [δ values for C-3 (25.5), C-4 (149.2), C-5 (76.3), and C-6 (36.8)] and difference from those of structurally similar compounds having a β -orientation of the hydroxyl group at C-5, (1*R*,5*R*,7*S*,9*S*,10*S*)-9-acetoxy-l(10)-epoxy-5-hydroxygermacra-4(15),11(13)-diene [δ values for C-3 (22.9), C-4 (147.3), C-5 (75.6), and C-6 (35.9)]. Moreover, the ROESY correlation between H $_{\beta}$ -5 (δ 3.88) and H-7 (δ 1.67) confirmed the same orientation of H-5 and H-7. The 13 C NMR chemical shifts from

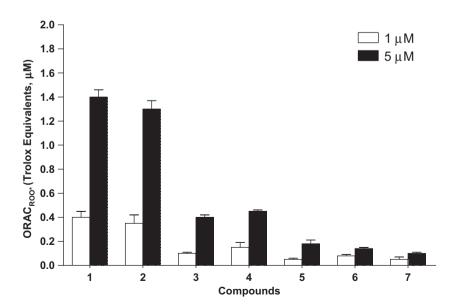


Figure 4. Peroxyl radical-scavenging activity (Trolox equivalent, μ M) of 1–7. The ORAC value is calculated by dividing the area under the sample curve by the area under the Trolox curve, with both areas being corrected by subtracting the area under the blank curve. One ORAC unit is assigned as the net area of protection provided by Trolox at a final concentration of 1 μ M. The area under the curve of the sample is compared to the area under the curve for Trolox, and the antioxidative value is expressed in micromoles of Trolox equivalent per liter. The results represent the mean \pm S.D. of values obtained from three measurements.

b Mitoxantrone (MX), an anticancer agent, was used as a reference compound.

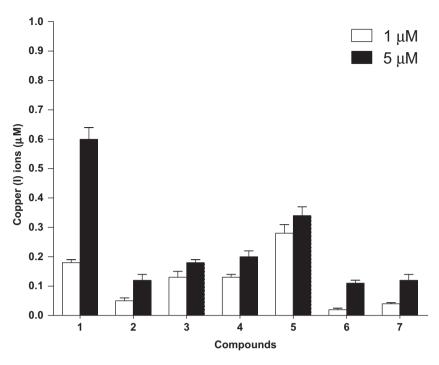


Figure 5. Reduction potential of 1-7. The results represent the mean ± S.D. of values obtained from three measurements.

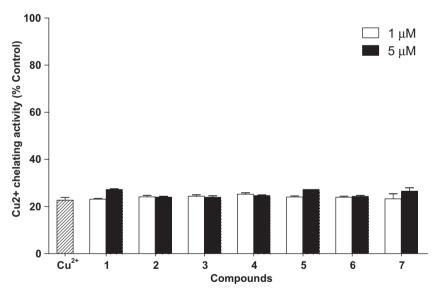


Figure 6. Metal chelating activity of 1-7. The results represent the mean ± S.D. of values obtained from three measurements.

C-13 to C-20 of **1** were identical to those of lobatriene (**4**), ^{3,4} indicating that these two compounds have the same B ring with a 2-hydroxyisopropyl group at C-17. The configuration at C-17 was determined to be 17*R* by excellent agreement of the ¹H and ¹³C NMR data of lobatriene⁴ and further confirmed by ROESY correlations between H-7 $_{\beta}$ (δ 1.67) and H $_{\beta}$ -16 (δ 2.13), H $_{\alpha}$ -16 (δ 1.93) and H $_{\alpha}$ -17 (3.25) (see Fig. 3). From all above evidence, the structure of **1** (Fig. 1) was elucidated and the compound was named lobocompactol A.

The molecular formula of lobocompactol B (**2**) was also $C_{20}H_{32}O_3$, as identified by ion peaks in the ESIMS at m/z 303 [M $-H_2O+H$]⁺ (positive) and FTICRMS at m/z 343.22449 [M+Na]⁺ (calcd for $C_{20}H_{32}O_3$ Na, 343.22491). Detailed analyses of the 1D and 2D NMR spectra indicated that **2** has the same planar structure as that of **1**. The chemical shifts of C-3 (δ 28.8), C-4 (δ 149.2), and

C-5 (δ 75.5) in the diterpene **2** confirmed hydroxyl group at C-5 to be α -orientation by comparing with the corresponding data of the gyrosanol C¹¹ [δ values for C-3 (25.5), C-4 (149.2), and C-5 (76.3)] and (1*R*,5*R*,75,95,10S)-9-acetoxy-l(10)-epoxy-5-hydroxygermacra-4(15),11(13)-diene¹² [δ values for C-3 (22.9), C-4 (147.3), and C-5 (75.6)]. However, there were some noticeable difference in the ¹³C NMR chemical shifts, especially C-6 (δ 32.5), C-7 (δ 37.1), and C-13 (δ 145.1), indicating the change of configuration at C-7 in **2** by comparing with that of **1**. This was further confirmed by none ROESY correlations between H_{β}-5 (δ 4.05) and H-7 (δ 2.12). The configuration at C-17 was also determined to be 17*R* by comparing ¹H and ¹³C NMR data for the B ring of **2** with those of lobatriene⁴ and further confirmed by ROESY correlations between H_{α}-16 (δ 1.97) and H_{α}-17 (δ 3.27) (see Fig. 3). Thus, the structure of **2** was confirmed and named lobocompactol B.

Compounds 1–7 were evaluated for cytotoxic activity against A549 (lung) and HL-60 (acute promyelocytic leukemia) human cancer cell lines, after continuous exposure for 72 h (Table 2). Compound **5** exhibited a strong activity against the A549 cell line with an IC₅₀ of 4.97 \pm 0.06 μ M, while **3**, **4**, and **7** exhibited moderate activity with IC₅₀ values of 23.03 \pm 0.76, 31.13 \pm 0.08, and 36.45 \pm 0.01 μ M, respectively. Compounds **1**, **2**, and **6** were inactive (IC₅₀ >100 μ M). The cytotoxic activity of **5** on the A549 cells was comparable to that of the positive control, mitoxantrone (MX). All compounds showed moderate cytotoxic activity against the HL-60 cell line with IC₅₀ values ranging from 17.80 \pm 1.43 to 59.06 \pm 2.31 μ M.

The antioxidant capacity of compounds **1–7** was measured using an oxygen radical absorbance capacity (ORAC) assay (see Fig. 4). Compounds **1** and **2** showed moderate peroxyl radical-scavenging activities of 1.4 and 1.3 μ M Trolox equivalents, respectively, at a concentration of 5 μ M. The others showed weak or no activity. All of the evaluated compounds exhibited low reducing potentials and weak metal chelating activity (see Figs. 5 and 6).

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Supplementary data

Supplementary data associated with this article can be found, in the online version, at doi:10.1016/j.bmcl.2011.01.072.

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